

**INFLUENCE OF TIME AND TEMPERATURE ON  
RESISTIVITY AND MICROSTRUCTURE OF  
 $\text{Cu}_x\text{Co}_{1-x}\text{Fe}_2\text{O}_4$  MIXED FERRITES**

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**Abstract**—Copper-cobalt ferrites with general chemical formula  $\text{Cu}_x\text{Co}_{1-x}\text{Fe}_2\text{O}_4$  (with  $x = 0.0, 0.4, 0.6$  &  $1.0$ ) were prepared by ceramic method. The solid state reaction was confirmed by XRD patterns. DC conductivity was measured by two probe method. Electrical resistivity is found to increase on lowering of sintering temperature and time. At  $x = 1.0$ , the conduction is mainly due to hopping of electrons leading to n-type conductivity while at  $x = 0.0$ , conduction is due to holes leading to P-type conductivity. The lowest conduction at  $x = 0.4$  is attributed to the electron hole compensation. SEM micrographs were obtained from JEOL scanning electron microscope. The micrographs reveal that an average grain size increases with sintering temperature and time as a result of decrease in porosity. This leads to the decrease in resistivity with sintering temperature and time. One of the factors for higher conductivity in ferrites is an increase in average grain size and decrease in pore concentration during the heat treatment.

## **1. INTRODUCTION**

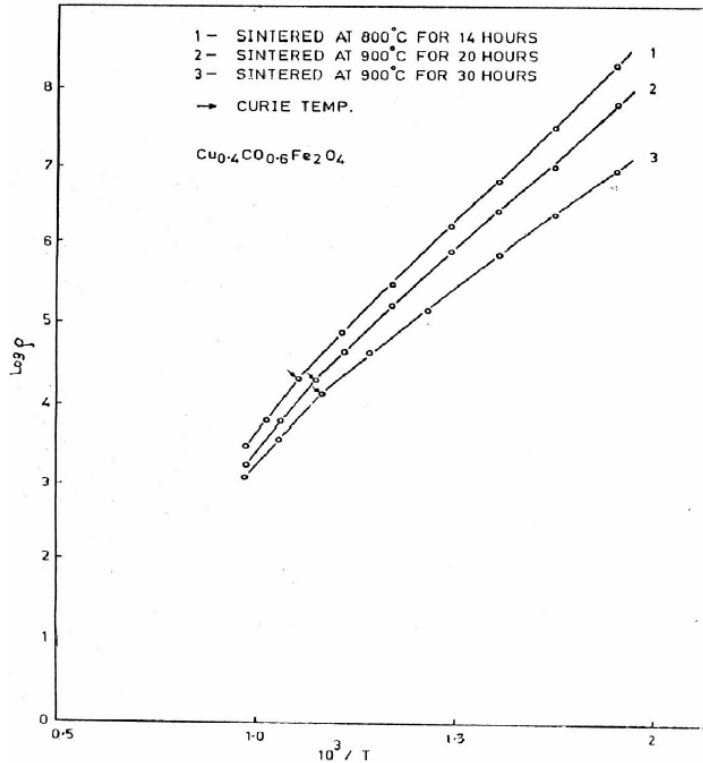
Ferrites have wide applications from microwave to radio frequencies [1, 2]. They possess very low conductivity, which is one of the considerations for microwave applications. The conductivity greatly

influences the dielectric [3–7] and magnetic behavior of ferrites [8,9]. This created considerable interest in many research workers for the development and potential applications of ferrites in the electronic industry [10–12]. Now a days the study of ferrites has occupied an important place in the electronics technology [13]. The practical applications of ferrites are now of immense service in every day life, as electronic and magnetic ceramics. Because of their poor electrical conductivity [14], they have revolutionized the field of high and ultrahigh frequency electronics with negligible eddy current losses. This is the reason why ferrites have occupied a unique position in this field.

Many parameters play an important role in determining a particular application of material. The quantities such as magnetization, coercivity, permittivity [15], conductivity etc. are greatly influenced by porosity, grain size and microstructure of the sample [16, 17]. Ferrites containing Cobalt and Copper exhibit several interesting properties which make them suitable for switching and memory devices. At frequencies above 50 MHz cobalt is preferred to preclude magnetic losses [18]. Cobalt is invariably doped to various ferrites for the rapid relaxation to lattice [19]. In this paper we report the influence of composition and heat treatment on resistivity and microstructure of  $\text{Cu}_x\text{Co}_{1-x}\text{Fe}_2\text{O}_4$  ferrites.

## 2. EXPERIMENTAL DETAILS:

The ferrites with general chemical formula  $\text{Cu}_x\text{Co}_{1-x}\text{Fe}_2\text{O}_4$  (with  $x = 0.0, 0.4, 0.6$  &  $1.0$ ) were prepared by conventional ceramic method using AR grade copper oxide, Cobalt oxide and Ferric oxide. The compositional weights of powders were mixed physically and blended in agate mortar in acetone medium. All the samples were presintered at about  $800^\circ\text{C}$  for 20 hrs. The presintered powders were subjected to hard milling process in acetone medium for 6 hrs. The powder was subjected to pellet formation; these samples in the pellet form were finally sintered at  $800^\circ\text{C}$  for 14 hrs,  $900^\circ\text{C}$  for 20 hrs and  $900^\circ\text{C}$  for 30 hrs. The solid state reaction was confirmed by XRD patterns and the absence of any extra line confirms the formation of single phase ferrites. DC conductivity was measured by a two probe method on compressed and sintered pellets. Micrographs were taken using JEOL scanning electron microscope.



**Figure 1.** Variation of resistivity with temperature.

### 3. RESULTS AND DISCUSSION:

The variation of resistivity as a function of temperature is shown in Figure 1. The change in slope is markedly observed in all the samples, such change is either due to Curie temperature [20] or due to change in conduction mechanism [21]. This indicates the semiconducting nature of ferrites. The resistivity in ferrites obeys the relation

$$\rho = \rho_0 \exp (\Delta E / K T)$$

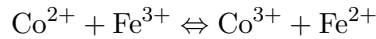
where  $\Delta E$  = Activation energy,  $K$  = Boltzmann constant &  $T$  = Absolute temperature. Each sample shows a break near the Curie temperature which is attributed to the transition from ferri to paramagnetic region. The activation energies are calculated from the slopes of the paramagnetic and ferromagnetic region and tabulated in the Table 1 along with sintering temperature, time and Curie temperature. From Table 1 it is evident that the lower activation

energy in the ferromagnetic region is attributed to the phase transition or impurity phases, while the change in activation energy is attributed to the change in conduction mechanism [22].

**Table 1.** Data on sintering temp/time, activation energy, Curie temperature and resistivity.

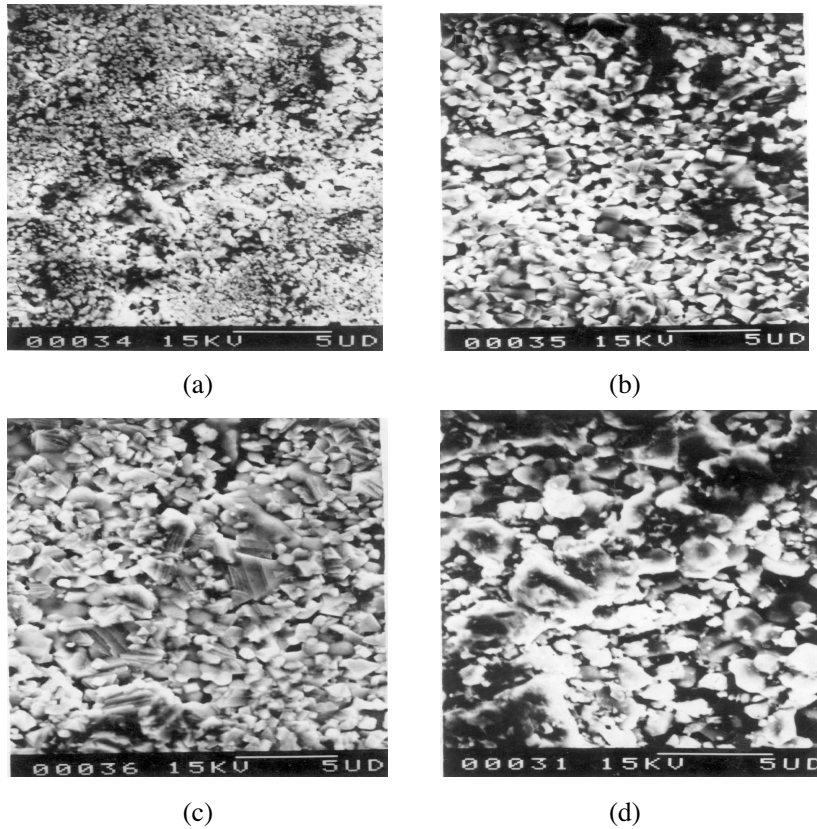
Sample	Sintering temp	Sintering time	Activation energy (eV)		Curie temp	Resistivity At 250°C
			Para-region	Ferri-region		
CoFe <sub>2</sub> O <sub>4</sub>	800 °C	14 hrs	1.72	0.49	590 °C	174 K.Ω-cm
	900 °C	20 hrs	1.03	0.48	545 °C	71 K.Ω-cm
	900 °C	30 hrs	0.93	0.45	520 °C	23 K.Ω-cm
Cu <sub>0.4</sub> Co <sub>0.6</sub> Fe <sub>2</sub> O <sub>4</sub>	800 °C	14 hrs	1.29	1.00	540 °C	274 M.Ω-cm
	900 °C	20 hrs	1.14	0.90	495 °C	93 M.Ω-cm
	900 °C	30 hrs	1.02	0.74	470 °C	10 M.Ω-cm
CuFe <sub>2</sub> O <sub>4</sub>	800 °C	14 hrs	1.59	0.45	520 °C	33 K.Ω-cm
	900 °C	20 hrs	1.34	0.36	480 °C	13 K.Ω-cm
	900 °C	30 hrs	0.97	0.34	450 °C	2 K.Ω-cm

The change in activation energy for different compositions is attributed to the hopping of polarons. The values of activation energy above 0.2 eV clearly indicate the polaron hopping in the system [23]. The addition of cobalt replaces Fe<sup>3+</sup> ions from B site which intern replaces Cu<sup>2+</sup> ions in A site. There fore the presence of cobalt on B site favors the conduction mechanism in Cu-Co ferrites by polaron hopping [23].



In CuFe<sub>2</sub>O<sub>4</sub> Ferrite, Fe<sup>2+</sup> concentration is appreciable and conduction is due to hopping of electrons from Fe<sup>2+</sup> → Fe<sup>3+</sup> leading to n-type conductivity with relatively low activation energy. Low conductivity in CoFe<sub>2</sub>O<sub>4</sub> is due to deficiency of iron leading to P-type conductivity with relatively large activation energy (Table 1). The lowest conduction in Cu<sub>0.4</sub>Co<sub>0.6</sub>Fe<sub>2</sub>O<sub>4</sub> with largest activation energy is attributed to electron-hole compensation. The decrease in Curie temperature with increase in sintering temp/time is attributed to the decrease in A-B interactions [24].

The SEM micrographs of the samples in the present case are as shown in Figure 2 (2(a) to 2(d)) along with the conditions of sintering for each samples. These micrographs reveal the following important features. The porosity of the samples varies from 20% to 30% as they have been prepared by ceramic method. Depending on the heat



**Figure 2.** SEM micrographs of  $\text{Cu}_x\text{Co}_{1-x}\text{Fe}_2\text{O}_4$  at different sintering temperature/time. (a)  $x = 0$  sintered at  $800^\circ\text{C}$  for 14 hrs, (b)  $x = 0$  sintered at  $900^\circ\text{C}$  for 20 hrs, (c)  $x = 0$  sintered at  $900^\circ\text{C}$  for 30 hrs, (d)  $x = 1$  sintered at  $800^\circ\text{C}$  for 14 hrs.

treatment the grain size varies upto  $10\ \mu\text{m}$ . Figure 2(c) shows the maximum grain size in a sample sintered at  $900^\circ\text{C}$  for 30 hrs.

The samples sintered at low temperature ( $800^\circ\text{C}$ ) for 14 hrs exhibit almost open pores along with relatively small grains. Thus concentration of porosity is found to decrease with increase in sintering temperature and time [25]. In case of exaggerated grain growth (Figure 2(c)), closed pores are trapped inside the grains. They are isolated from the grain boundaries and hence cannot shrink any more. Under such condition sintering is practically inhibited. A similar behaviour has been reported in case of Mg-Mn-Zn ferrites [26]. The mechanism of pore growth combined with grain growth results in

the characteristic microstructure of ferrites in which residual porosity appears in intragranular space. This is clearly indicated in the samples (Figures 2(a) to 2(d)).

The close survey of the micrographs reveals that the samples sintered at 800 °C for 20 hrs contain large number of small grains with more porosity than the samples of same composition sintered at 900 °C for 20 hrs. The micrographs also represent two typical microstructures that cover entire field of magneto ceramics. One type has a high density containing large grains (Figure 2(c)) more than 10  $\mu\text{m}$  and remaining porosity is present as clouds of fine spherical pores inside the crystallites. Such microstructure may due to oxygen vacancies as a result of reduction during firing. Another type consists of smaller and pore free crystallites (Figures 2(a) & 2(d)) however the large pores are present at the boundaries of crystallites. Such structure is due to the certain amount of metal ion vacancies caused by oxidation or suitable doping. These results are in good agreement with those reported for Ni-Zn ferrites [26].

It can be seen from the micrographs that the average grain size increases with increase in temperature/time with decrease in porosity [27]. This results higher conductivity in the samples as a consequence of decrease in resistivity [28]. The presence of air gaps between the grains result in the formation of inhomogeneous dielectric structure. This greatly affects the DC and AC resistivity of ferrites and hence conduction mechanism in ferrites is largely dependent on porosity. Therefore smaller the porosity greater will be the grain size and higher will be the conductivity. The micrographs indicate the low porosity in a matrix of large sized grains in ferrites sintered at higher temperature for a longer time. This causes decrease in resistivity and increase in conductivity. Therefore it can be emphasized that one of the factor for higher conductivity in ferrites is the increase in average grain size and decrease in pore concentration during the heat treatment.

#### 4. CONCLUSION

In the present study, we have investigated the effect of sintering temperature and time on resistivity and microstructure of  $\text{CuCoFe}_2\text{O}_4$  ferrites. The increase in electrical conductivity as a function of sintering temperature/time is due to increase in grain size. The average grain size increases with sintering temperature and time as a result of decrease in porosity. This intern decreases the resistivity with sintering temperature and time. There fore it is concluded that sintering temperature and time plays dominant role in deciding the electrical conductivity and microstructure of the particular ferrite.

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