

Two-Photon Luminescence and Second Harmonic Generation of Single Layer Molybdenum Disulphide Nanoprobe for Nonbleaching and Nonblinking Optical Bioimaging

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Abstract—Layered molybdenum disulphide (MoS₂) can efficiently emit photoluminescence (PL) excited by visible light. However, one-photon PL of MoS₂ for bioimaging purposes suffers from strong autofluorescence and ion-induced PL quenching. Herein, we report single layer chitosan decorated MoS₂ nanosheets as nonbleaching and nonblinking optical nanoprobe under near infrared femtosecond laser excitation and their applications for two photon luminescence (TPL) and second harmonic generation (SHG) bioimaging. The TPL can resist the ion-induced quenching by the cellular membrane. The proposed TPL and SHG of single layer MoS₂ show great potential for real-time, deep and multiphoton bioimaging.

1. INTRODUCTION

Layered two-dimensional nanomaterials have attracted much attention since the birth of graphene isolated from graphite [1]. Beyond graphene, transition-metal dichalcogenides (TMDs) have been highly studied because of their rich electronic, optical, mechanical and chemical properties as well as promising applications in sensing, catalysis, energy storage and biomedicine [2–5]. Molybdenum disulphide (MoS₂), containing of one plane of molybdenum atoms sandwiched between two planes of sulfur atoms, is a primary focused TMD [6–8]. Because of the relatively weak Van der Waals forces between their layers and the strong intralayer interactions, the exfoliation of single layer MoS₂ from its bulk counterpart was realized [9]. Unlike the bulk materials, single layer MoS₂ nanosheets (SLMNs) are a direct band gap semiconductor material, making them emit strong photoluminescence (PL) [10, 11]. The one-photon excited luminescence from layered MoS₂ nanosheets has already been applied to biosensing and bioimaging research (although the PL of multilayer MoS₂ is relatively quite weak) [12, 13].

However, the ultraviolet-visible light excited one-photon luminescence bioimaging applications has some drawbacks, such as strong autofluorescence and the very limited imaging depth due to the strong scattering of short-wavelength photons [14]. Nonlinear optical imaging using near infrared (NIR) light can avoid those drawbacks. Nonlinear optical imaging (e.g., two-photon luminescence (TPL) and second harmonic generation (SHG) imaging) using the excitation wavelength in the first NIR window (650–900 nm) or the second NIR window (1000–1300 nm) has many advantages, such as low autofluorescence, reduced photobleaching and the remarkable capability for deep tissue imaging [15–17]. Although the extinction coefficient of SLMNs is larger than those of graphene oxide and gold nanorods under the NIR [18, 19], the one photon excited PL is hard to be detected at the long (e.g., 600 nm) excitation wavelength. Recently, an exceedingly large two-photon absorption coefficient (larger than that of

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conventional semiconductors by three orders of magnitude) and a large laser damage threshold of single layer MoS₂ island were reported [20]. SHG and third harmonic generation from single or few layers MoS₂ film were also reported [21–23]. MoS₂ nanosheets are expected to be NIR-excited optical nanoprobes for bioimaging. However, we are hardly aware of research on the TPL and SHG properties of small size (sub-100 nm) SLMNs and their application in bioimaging.

In this work, we reported the strong and stable TPL and SHG from small chitosan (CS) decorated SLMNs (CS-MoS₂ nanosheets) with sub-100 nm lateral dimension. Excited by the first NIR window light, CS-MoS₂ nanosheets show strong TPL with its peak at 525 nm at a low power of 3 mW. Under the excitation of the second NIR window, strong SHG with sharp peaks were observed at a lower power of 1 mW. In addition, the TPL and SHG from CS-MoS₂ nanosheets were demonstrated to be nonbleaching and nonblinking. As a result, CS-MoS₂ nanosheets were carried out to TPL and SHG imaging and cellular three-dimensional (3D) scanning imaging under low power laser excitation. Furthermore, the reported quenching of TPL from MoS₂ nanosheets on the cellular membrane was not observed, indicating their feasibility in bioimaging.

2. EXPERIMENTAL

2.1. Materials

N,N-Dimethylformamide (DMF), acetone, methanol and acetic acid were purchased from Sino Chemical Reagent Co., Ltd. Ionic liquid (IL) 1-butyl-3-methylimidazolium hexafluorophosphate (BMIPF₆) ($\geq 97.0\%$) and chitosan (CS) were purchased from Aladdin Reagent. Molybdenum sulfide (MoS₂) with a bulk particle size $< 2 \mu\text{m}$ were purchased from Sigma-Aldrich. Cell Counting Kit-8 (CCK-8, 99.9%) was purchased from Beyotime Biotechnology. Dulbecco's modified eagle medium (DMEM, 99%), phosphate buffered saline (PBS), and fetal bovine serum (FBS, 99%) were purchased from Gioco Life technology. All chemical reagents were used without further purification, and deionized water was used in all the experimental procedures.

2.2. IL-Assisted Exfoliation of Single Layer CS-MoS₂ Nanosheets

The preparation of CS-MoS₂ nanosheets was done according to Zhang's method with some modifications [24]. 250 mg of bulk MoS₂ powders and 0.5 mL of IL were added to the mortar and ground forcibly for 90 minutes. Then 100 mg of CS was then added into the mortar, followed by grinding for another 30 minutes. The collected mixture was added into a solution of DMF and acetone (DMF : acetone = 1 : 1, v/v) and centrifuged at 12000 rpm for 20 minutes. In order to remove the excess CS, the mixture was added into acetic acid solution (acetic acid : water = 0.5%, v/v) and centrifuged at 12000 rpm for 20 minutes. The entire washing cycle was repeated twice. The collected precipitant was re-dispersed into deionized water and further centrifuged at 2000 rpm for 45 min, twice. The supernatant containing CS-MoS₂ nanosheets was diluted with water to 5 mL for the following investigations. The resulting concentration of CS-MoS₂ nanosheets in water was calculated to be 1.15 mg/mL by drying the supernatant. The MoS₂ nanosheets and CS-MoS₂ nanosheets utilized the same purification method.

2.3. Characterization

Transmission electron microscope (TEM) images were obtained with a JEM-2100HR transmission electron microscope (JEOL). Atomic force microscopy (AFM) images were obtained with a Bruker Multimode 8 atomic force microscope. The absorption spectra of the CS-MoS₂ nanosheets in deionized water were acquired with a Lambda 950 UV-vis-NIR spectrophotometer (PerkinElmer). Photoluminescence spectra and images of CS-MoS₂ nanosheets were recorded using a home-built optical measurement system.

2.4. Optical Measurement System

The optical measurement system was based on an inverted multi-photon scanning microscope (IX81, Olympus), a tunable Ti: Sapphire laser as the excitation source (Coherent Mira HP, 76 MHz femtosecond

pulse mode at 750–920 nm), an optical parameter oscillator (Synchronized Coherent Mira OPO, 76 MHz femtosecond plus mode 1000–1600 nm), and a fibre spectrometer (QE65Pro, Ocean Optics). The excitation beam from the Mira HP or Mira OPO passed through a polarizing beamsplitter cube (PBS, Thorlabs) and a galvanometer system. After being reflected by a 670-nm dichroic mirror (Semrock) or 950-nm dichroic mirror (Chroma) and directed into the oil immersion objective (60 \times , NA = 1.35, Olympus), the excitation laser beam was focused on the sample. The luminescence signal from the sample was backward captured by the same objective. A band-pass filter (FF01-525/30-25 or FF01-550/49-25, Semrock) was used to further block the excitation light between objective and reflection mirrors. The fluorescence light was directed into a PMT to obtain fluorescence images (Excitation power: 0–5 mW). In addition, without silver mirror (M5), the fluorescence light was collected by a fluorescence spectrometer for spectral analysis.

2.5. Cytotoxicity of CS-MoS₂ Nanosheets

HeLa cells (6×10^3) in 100 μ L of DMEM medium (10% FBS, 100 U/mL penicillin, 100 μ g/mL streptomycin) were seeded onto 96-well plates for 24 hours. The cells were then incubated with different concentrations of CS-MoS₂ nanosheets for 12 hours. To remove the redundant nanosheets, the cells were washed by PBS twice and cultured in a flash medium for another 24 hours. Afterward, 10 μ L of the CCK-8 solution were added to each well, followed by incubation for another 1 hour. Cell viability was determined by measuring the absorbance at the wavelength of 450 nm with a microplate reader. All measurements were conducted in triplicate.

2.6. Bioimaging of CS-MoS₂

Hela cells were seeded onto the confocal dish and grown in 1.5 mL of complete medium (DMEM medium, 10% FBS, 100 U/mL penicillin, 100 μ g/mL streptomycin) at 37°. After 24 hours, the cells were incubated with CS-MoS₂ nanosheets for 12 hours. The excess nanosheets were then washed by PBS, twice. The cells were observed via scanning microscopy with a 60 \times oil immersion objective lens with excitation at 750 nm and 1050 nm.

3. RESULTS AND DISCUSSION

SLMNs were exfoliated from bulk MoS₂ powders and then transferred to become soluble according to previously reported method with some modifications [24]. The atomic force microscopy (AFM) image (Figure 1(a)) shows that the thickness of the exfoliated MoS₂ nanosheets is 0.7–1 nm, indicating the single layer structure (0.65 nm in theory) [25]. To improve the chemical stability and biocompatibility, SLMNs were decorated with CS [26]. The presence of CS in the CS-MoS₂ nanosheets is characterized by the results of Fourier transform infrared spectra (Figure 1(b)). The size of the CS-MoS₂ nanosheets were determined to be below 100 nm (Figure 1(c)). Compared to SLMNs, the thickness of CS-MoS₂ nanosheets increase to \sim 1.5 nm due to the attachment of CS. The formation of single-layered structures of CS-MoS₂ nanosheets was also confirmed by the observation of no multi-steps on the edges (Figure 1(d), the multi-layer CS-MoS₂ nanosheets were obtained from the precipitant of the final process.) [27]. Raman spectra of CS-MoS₂ nanosheets were measured to give bands at 384 cm^{-1} (the in-plane vibration E_{2g}^1 mode of 2H-MoS₂ crystal) and 407 cm^{-1} (the out-of-plane vibration A_{1g} mode of 2H-MoS₂ crystal), respectively (Figure 1(e)) [28]. Being exfoliated from bulk powders, the wavenumber difference of CS-MoS₂ nanosheets between the two peaks decreases from 25.5 to 22.5 cm^{-1} as the number of layers decreases because of the anomalous lattice vibration [29]. It is worth mentioning that the liquid-phase exfoliated single layer nanosheets showed a different position from the mechanically exfoliated single layer nanosheets due to surface adsorption [9].

CS-MoS₂ nanosheets show four characteristic optical absorption bands with peaks at 400 nm, 447 nm, 606 nm and 667 nm, in good agreement with the observation of the liquid exfoliated SLMNs (Figure 2(a)) [9]. The relatively weak optical absorption bands at 606 nm and 667 nm and the strong absorption peaks at 400 nm and 447 nm suggest the existence of a large number of small sized nanosheets due to the quantum size effect [30]. In general, layered MoS₂ with large size shows two down-converted

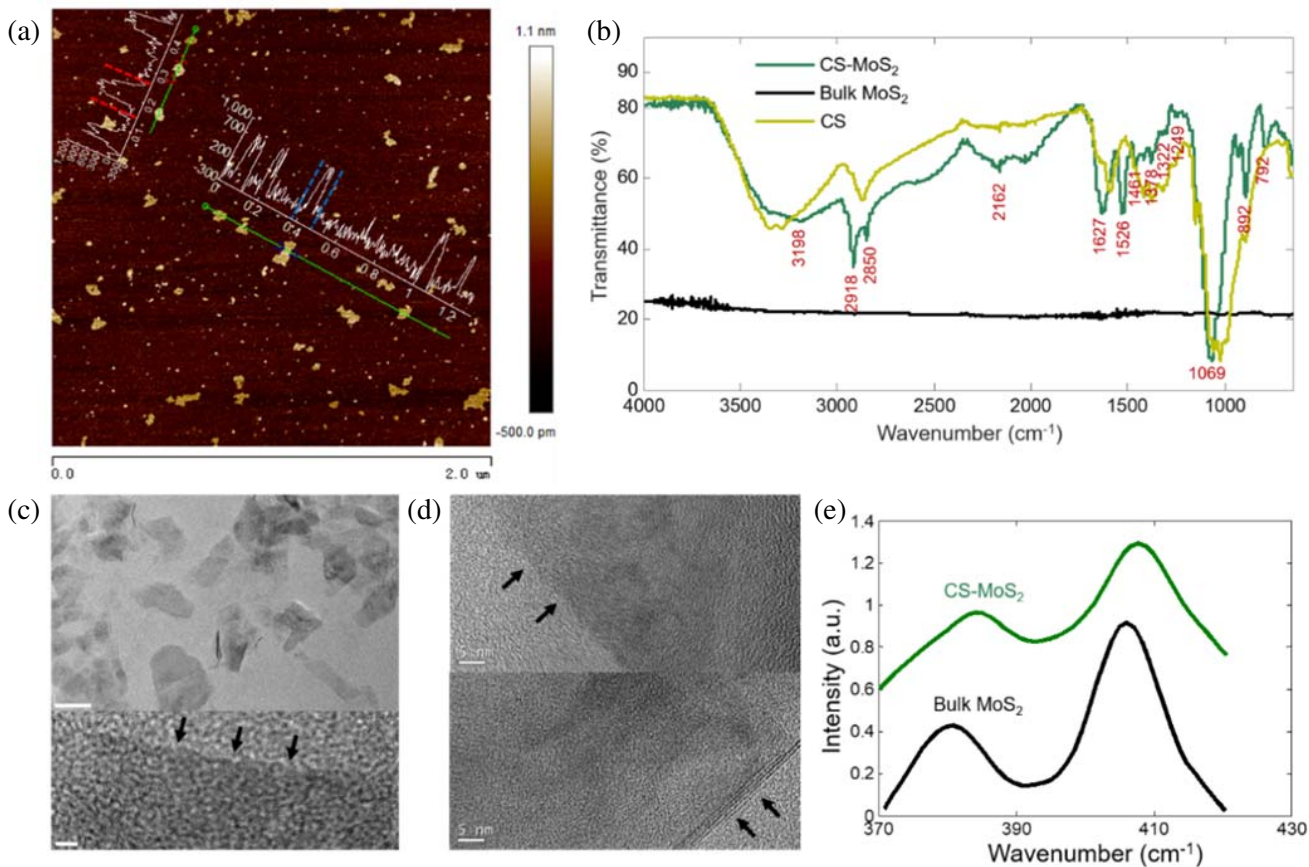


Figure 1. (a) AFM image of SLMNs. Inset: Thickness profile of SLMNs along with green lines. The unit of values on the vertical axis are picometers. (b) Fourier transform infrared spectra for CS, bulk MoS₂ and CS-MoS₂ nanosheets. (c) Upper: TEM image of CS-MoS₂ nanosheets. The scale bar: 50 nm. Bottom: High-resolution TEM image at an edge (indicated by black arrows) of a CS-MoS₂ nanosheet. The scale bar: 2 nm. (d) TEM images of CS-MoS₂ nanosheets at edges. Upper: single layer CS-MoS₂ nanosheets. Bottom: 4–5 layers CS-MoS₂ nanosheets. The scale bar: 5 nm. (e) Raman spectra of bulk MoS₂ and CS-MoS₂ nanosheets. It should be noted that there is no obvious Raman peaks from chitosan due to richness of functional groups such as -CH₂, -COOH, -NH₂, whose Raman peaks are mainly located between 1000 and 4000 cm⁻¹.

PL peaks at around 627 nm and 670 nm when excited by continuous visible light [31]. When the lateral dimension of the layered MoS₂ decreases to 100 nm or smaller, these two peaks are sharply weakened and another new blue-shifted PL band appears. The new peak is excitation-wavelength dependent and can be attributed to the blue-shifted hot PL from the K point [12, 32]. This phenomenon would induce cross-talk among emission detection channels in fluorescent microscopic imaging. In addition, the emission of the two peaks cannot be observed at the long (e.g., 600 nm) excitation wavelength [32]. Short wavelength excitation would impose the problems of strong autofluorescence in bio-samples as well as the resulting limited imaging depth. Therefore, one photon excited PL of layered MoS₂ nanosheets is not suitable for bioimaging applications. In contrast, nonlinear optical imaging is more suitable for bioimaging. Herein, the femtosecond (fs) laser was used to give a comprehensive view of the TPL and SHG properties of CS-MoS₂ nanosheets. As shown in Figure 2(b), excited by the laser of 750–920 nm, the PL peak of CS-MoS₂ nanosheets is unvarying and centered at 525 nm. A 1.94 order of excitation power dependence of PL was obtained (Figure 2(c)), indicating that the PL is due to two-photon excitation. Interestingly, unlike one photon excited luminescence, the TPL spectra of the layered MoS₂ nanosheets were not found to be excitation-wavelength dependent. This allows great flexibility in choosing the light source. Since

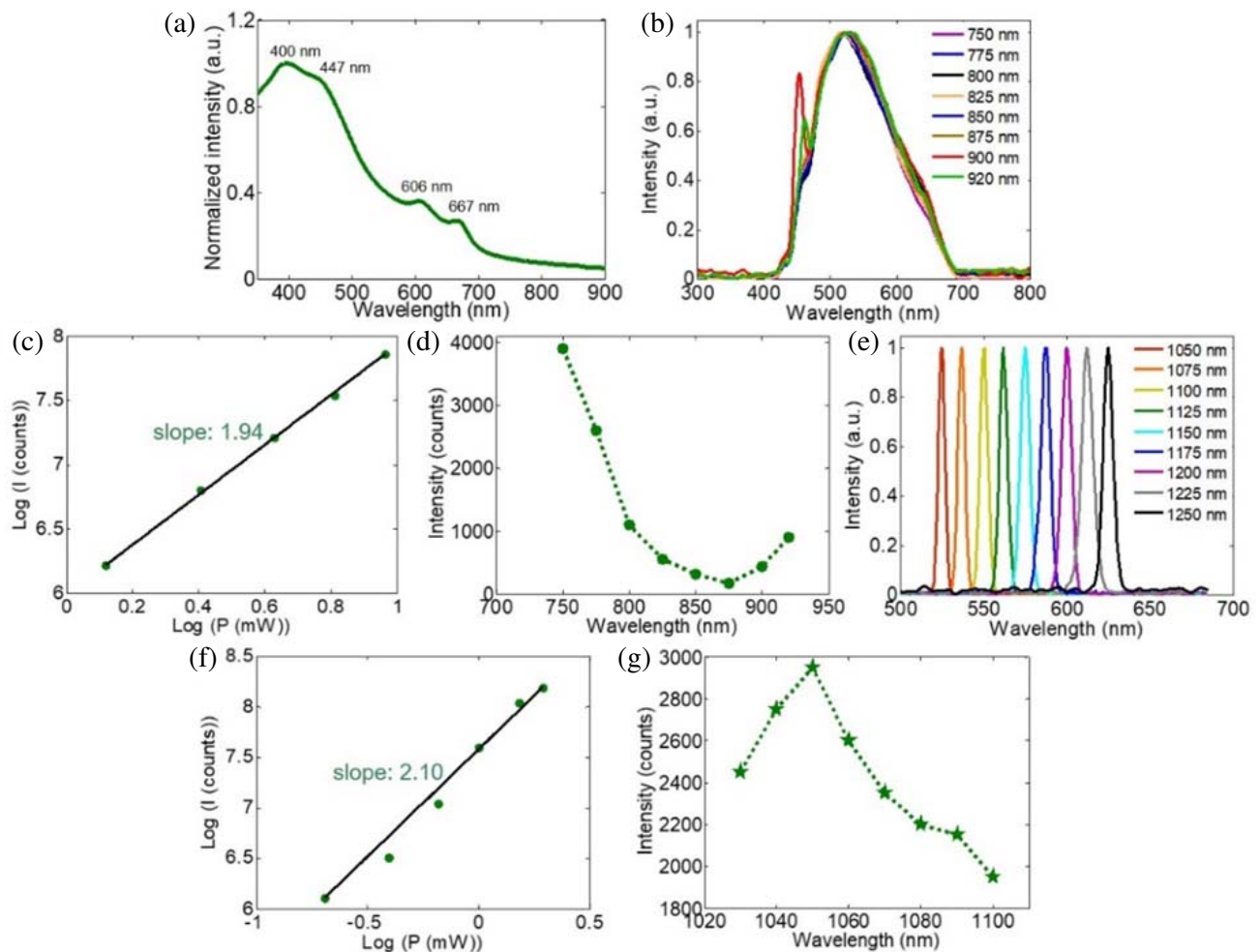


Figure 2. (a) Absorption spectra of CS-MoS₂ nanosheets. (b) TPL spectra of CS-MoS₂ nanosheets excited by a range of wavelengths from 750 nm to 920 nm. (c) A logarithmic plot of the TPL intensity as a function of excitation power. (d) Related TPL intensity of CS-MoS₂ nanosheets under the different excitation wavelengths with the same power. (e) SHG spectra of CS-MoS₂ nanosheets excited by a range of wavelengths from 1050 nm to 1250 nm using micro-spectroscopy. The spectral resolution is about 1 nm. (f) A logarithmic plot of the SHG intensity as a function of excitation power. (g) Related SHG intensity of CS-MoS₂ nanosheets under the different excitation wavelengths with the same power.

the emission wavelength of the one photon excited PL is a function of the lateral dimension of MoS₂ nanosheets, we speculate that this phenomenon may result from the relatively even distribution of their lateral dimensions [12]. When varying the excitation wavelength at the same power (3 mW), 750 nm light exerts the best excitation efficiency (Figure 2(d)). Interestingly, two additional sharp peaks were observed when CS-MoS₂ nanosheets are excited by 900 nm and 920 nm light. According to the frequency relationship between the incident light and the emergent light peak in the spectra, they are potentially attributable to the SHG. These signals have not been detected under the excitation of 750–875 nm, probably because the intensity of TPL is much larger than that of SHG as well as the limitations of the optical detection system setup. This phenomenon also appeared when excited by the wavelength at the second NIR optical window. The spectral profiles of CS-MoS₂ nanosheets were observed clearly when CS-MoS₂ nanosheets were excited by 1050–1250 nm light, as shown in Figure 2(e). A 2.1 order of excitation power dependence of luminescence was obtained (Figure 2(f)), in good agreement with the second order nonlinearity of the SHG process. Under the illumination of different wavelength (1030–1100 nm) with the same power (1 mW), 1050 nm light shows the greatest efficiency (Figure 2(g)). No

TPL were observed under the excitation of 1030–1250 nm, probably due to the wavelength dependent TPL efficiencies. The CS-MoS₂ nanosheets showed strong nonlinear optical signals under the excitation of both the first and second NIR windows with low power, indicating their great potential for optical bioimaging applications.

Most organic dyes suffer from serious photobleaching effects. Semiconductor Quantum dots have very high emission efficiency and have no photobleaching, but rather suffer from photoblinking [33]. These are not beneficial to optical bioimaging, and it is necessary to develop optical nanoprobe possessing greater photostability properties. The photostability properties of TPL and SHG of CS-MoS₂ nanosheets were explored under continuous fs laser exposure for an extended time. The TPL intensity of CS-MoS₂ nanosheets have hardly any change under continuous fs laser exposure for an extended time of 600 seconds (Figure 3(a)), suggesting they do not suffer from photobleaching. To further explore whether the nonlinear optical signals of CS-MoS₂ nanosheets possess nonbleaching or nonblinking properties, the time traces of TPL and SHG intensities were obtained under continuous fs laser illumination. No TPL fluctuation is observed after 10 minutes of continuous scanning with 200 μ s temporal resolution (Figure 3(b)). The intensities of TPL coincide with a Gaussian distribution function (Figure 3(c)), indicating that CS-MoS₂ nanosheets are completely nonphotobleaching [34, 35]. Increasing the temporal resolution to 2 μ s, the nonphotoblinking property of CS-MoS₂ nanosheets was also demonstrated (Figures 3(d) and 3(e)) [36]. Using the same procedure, the nonbleaching and nonblinking properties of SHG from CS-MoS₂ nanosheets were also demonstrated. The absence of photobleaching and photoblinking effects, combined with the high brightness under NIR excitation, make CS-MoS₂ nanosheets a new type of optical probes for efficient imaging.

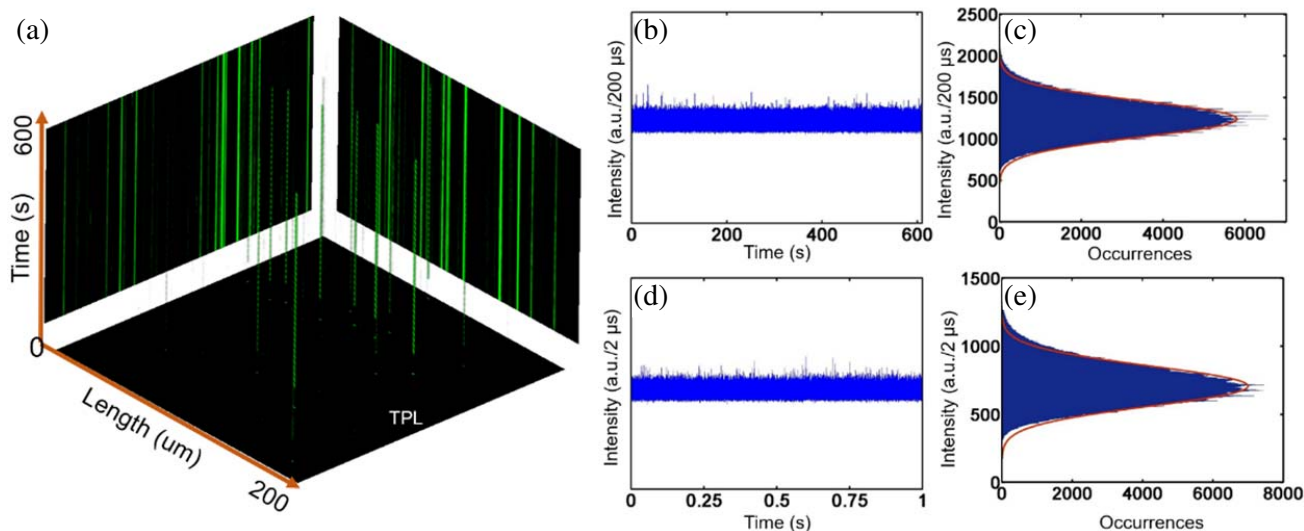


Figure 3. Photostability of TPL from CS-MoS₂ nanosheets. (a) Time trace repeated scanning image under continuous fs laser exposure for an extended time of 600 seconds. The pixel dwell time was 8 μ s. (b) The time trace of TPL intensity from a single light-emit spot under continuous fs laser illumination for 10 minutes with 200 μ s temporal resolution. (c) A histogram of TPL intensity of Figure 3(b) fitted with a Gaussian distribution function. (d) The time trace of TPL intensity from a single light-emit spot under continuous fs laser illumination for one second with 2 μ s temporal resolution. (e) A histogram of TPL intensity of Figure 3(d) fitted with a Gaussian distribution function.

Based on the above results, water-soluble CS-MoS₂ nanosheets promise to be stable and efficient optical nanoprobe for bioimaging. Herein, strong SHG signals were observed from the CS-MoS₂ nanosheet-treated HeLa cells under 1050 nm fs laser excitation (Figure 4(a)). TPL was observed with 750 nm fs laser (Figure 4(b)), indicating their potential for cell imaging. Furthermore, three-dimensional (3D) imaging of HeLa cells labeled with nanosheets exhibits the shell-like distribution of the TPL, as shown in Figure 4(c), suggesting that the cell membranes are well stained with the nanosheets.

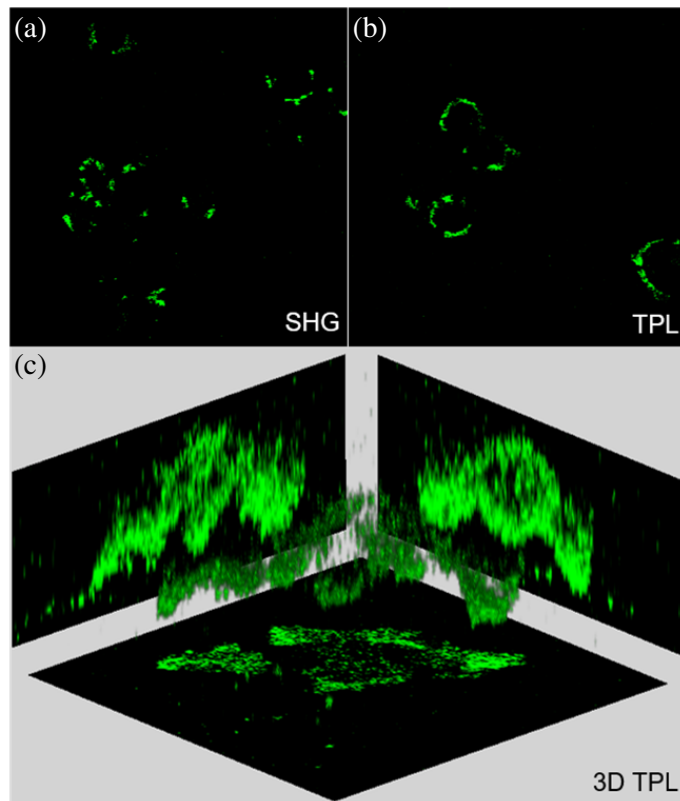


Figure 4. (a) SHG image of CS-MoS₂ nanosheet-treated HeLa cells. (b) TPL image of CS-MoS₂ nanosheet-treated HeLa cells. (c) 3D TPL image of CS-MoS₂ nanosheet-treated HeLa cells by Z-scan.

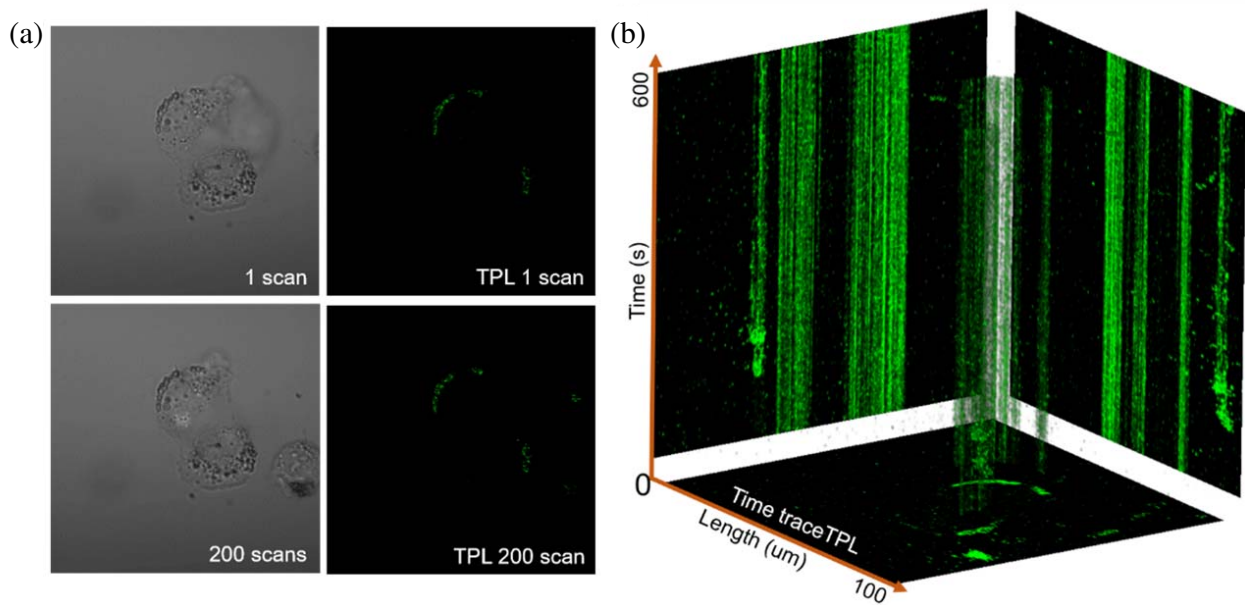


Figure 5. (a) The transmission and TPL images of CS-MoS₂ nanosheets-treated HeLa cells after the addition of methanol during continuous scanning. Each scan consumes ~ 3 seconds. (b) Time trace TPL of CS-MoS₂ nanosheets labeled in HeLa cells under continuous fs laser exposure for an extended time of 600 seconds. The pixel dwell time was $8 \mu\text{s}$.

Therefore, CS-MoS₂ nanosheets possess high imaging capability under the first and the second optical imaging window light excitation. As a result, CS-MoS₂ nanosheets reveal good potential for deep bioimaging applications.

Recently, some researchers indicated that the PL of layered MoS₂ nanosheets suffered from quenching on the cellular membrane [12]. The reason for the quenching phenomenon is the transformation of the crystal structure due to the intercalations of alkaline and H⁺ ions driven by the so-called membrane potential. However, this was not observed in this case. As shown in Figure 4, TPL from the CS-MoS₂ nanosheet-treated viable cells have not quenched during the scanning process. Figure 5(a) showed the scanning image of CS-MoS₂ nanosheet-treated cells after the addition of methanol. The cells have been nonviable and the intercalations of ions have been disabled. TPL intensities of CS-MoS₂ nanosheets on HeLa cells have hardly any increase between the first and the 200 times scanning images. The time trace TPL image with continuous fs laser exposure for an extend time of 600 seconds also showed hardly any increase in brightness (Figure 5(b)). There is no PL recovery occurred, suggesting that the ions on the cellular membrane have no effect on the TPL of CS-MoS₂ nanosheets. Two potential reasons may be related: (1) the decoration of CS on the surface of MoS₂ nanosheets will not only prevent the direct contact between the nanosheet and cell membrane, but also inhibit the intercalations of ions due to the electrostatic repulsion; (2) those ions cannot affect the PL intensity of single layer MoS₂ nanosheets. Therefore, the non-quenching property of CS-MoS₂ nanosheets improves their potential for bioimaging applications.

4. CONCLUSIONS

In conclusion, the TPL and SHG properties of single layer CS-MoS₂ nanosheets have been systematically studied. Excited by the laser light of first NIR optical window, CS-MoS₂ nanosheets emit strong TPL with a 525 nm band. Under the excitation of the second optical window, CS-MoS₂ nanosheets produce strong SHG. Both the TPL and SHG of CS-MoS₂ nanosheets have shown nonbleaching and nonblinking properties. As a result, CS-MoS₂ nanosheets have been used for TPL, SHG imaging and cellular 3D scanning imaging. Superior to the reported one-photon emission of MoS₂ nanosheets, the TPL of CS-MoS₂ nanosheets has been demonstrated to be non-quenching on the cellular membrane, adding to their advantages in bioimaging applications. The long wavelength NIR excited TPL and SHG from CS-MoS₂ nanosheets using low power also have great potential for deep tissue imaging.

Note

The authors declare that the files (incl. text, figures and data) related to this work leaked out somehow and consequently this work of ours was published by other people in *J. Biophotonics* in 2018, which has already been retracted after the fact was clarified that this work was originally conducted by Prof. Sailing He and his colleagues at South China Normal University, Guangzhou of China.

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